

Name _____ Section _____

Partner _____

CHM 111 Lab 3 Report Form

Write the balanced equation for this titration. Circle the analyte.

As always, attach one sample of each type of calculation. Watch sig figs!

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5
Mass pure KHP (g)					
Moles pure KHP					
Volume of NaOH used (mL)					
M NaOH					

Average M _____

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5
Deviation					

Average deviation _____

Be sure to indicate if any data were omitted from the calculations and your reasoning in leaving them out.