

Multiple Choice

| | | | | | | | | | |
|---|---------|----|---------|----|---------|----|---------|----|---------|
| 1 | A B C D | 9 | A B C D | 17 | A B C D | 25 | A B C D | 33 | A B C D |
| 2 | A B C D | 10 | A B C D | 18 | A B C D | 26 | A B C D | 34 | A B C D |
| 3 | A B C D | 11 | A B C D | 19 | A B C D | 27 | A B C D | 35 | A B C D |
| 4 | A B C D | 12 | A B C D | 20 | A B C D | 28 | A B C D | 36 | A B C D |
| 5 | A B C D | 13 | A B C D | 21 | A B C D | 29 | A B C D | 37 | A B C D |
| 6 | A B C D | 14 | A B C D | 22 | A B C D | 30 | A B C D | 38 | A B C D |
| 7 | A B C D | 15 | A B C D | 23 | A B C D | 31 | A B C D | 39 | A B C D |
| 8 | A B C D | 16 | A B C D | 24 | A B C D | 32 | A B C D | 40 | A B C D |

Numerical Response

| | | | | | |
|---------|---------|---------|---------|---------|---------|
| 1 | 2 | 3 | 4 | 5 | 6 |
| 0 0 0 0 | 0 0 0 0 | 0 0 0 0 | 0 0 0 0 | 0 0 0 0 | 0 0 0 0 |
| 1 1 1 1 | 1 1 1 1 | 1 1 1 1 | 1 1 1 1 | 1 1 1 1 | 1 1 1 1 |
| 2 2 2 2 | 2 2 2 2 | 2 2 2 2 | 2 2 2 2 | 2 2 2 2 | 2 2 2 2 |
| 3 3 3 3 | 3 3 3 3 | 3 3 3 3 | 3 3 3 3 | 3 3 3 3 | 3 3 3 3 |
| 4 4 4 4 | 4 4 4 4 | 4 4 4 4 | 4 4 4 4 | 4 4 4 4 | 4 4 4 4 |
| 5 5 5 5 | 5 5 5 5 | 5 5 5 5 | 5 5 5 5 | 5 5 5 5 | 5 5 5 5 |
| 6 6 6 6 | 6 6 6 6 | 6 6 6 6 | 6 6 6 6 | 6 6 6 6 | 6 6 6 6 |
| 7 7 7 7 | 7 7 7 7 | 7 7 7 7 | 7 7 7 7 | 7 7 7 7 | 7 7 7 7 |
| 8 8 8 8 | 8 8 8 8 | 8 8 8 8 | 8 8 8 8 | 8 8 8 8 | 8 8 8 8 |
| 9 9 9 9 | 9 9 9 9 | 9 9 9 9 | 9 9 9 9 | 9 9 9 9 | 9 9 9 9 |

| | | | | | |
|---------|---------|---------|---------|---------|---------|
| 7 | 8 | 9 | 10 | 11 | 12 |
| 0 0 0 0 | 0 0 0 0 | 0 0 0 0 | 0 0 0 0 | 0 0 0 0 | 0 0 0 0 |
| 1 1 1 1 | 1 1 1 1 | 1 1 1 1 | 1 1 1 1 | 1 1 1 1 | 1 1 1 1 |
| 2 2 2 2 | 2 2 2 2 | 2 2 2 2 | 2 2 2 2 | 2 2 2 2 | 2 2 2 2 |
| 3 3 3 3 | 3 3 3 3 | 3 3 3 3 | 3 3 3 3 | 3 3 3 3 | 3 3 3 3 |
| 4 4 4 4 | 4 4 4 4 | 4 4 4 4 | 4 4 4 4 | 4 4 4 4 | 4 4 4 4 |
| 5 5 5 5 | 5 5 5 5 | 5 5 5 5 | 5 5 5 5 | 5 5 5 5 | 5 5 5 5 |
| 6 6 6 6 | 6 6 6 6 | 6 6 6 6 | 6 6 6 6 | 6 6 6 6 | 6 6 6 6 |
| 7 7 7 7 | 7 7 7 7 | 7 7 7 7 | 7 7 7 7 | 7 7 7 7 | 7 7 7 7 |
| 8 8 8 8 | 8 8 8 8 | 8 8 8 8 | 8 8 8 8 | 8 8 8 8 | 8 8 8 8 |
| 9 9 9 9 | 9 9 9 9 | 9 9 9 9 | 9 9 9 9 | 9 9 9 9 | 9 9 9 9 |

PHYSICS DATA SHEET

Constants

| | |
|---|---|
| Acceleration Due to Gravity Near Earth..... | $ \vec{a}_g = 9.81 \text{ m/s}^2$ |
| Gravitational Constant | $G = 6.67 \times 10^{-11} \text{ N}\cdot\text{m}^2/\text{kg}^2$ |
| Radius of Earth | $r_e = 6.37 \times 10^6 \text{ m}$ |
| Mass of Earth..... | $M_e = 5.97 \times 10^{24} \text{ kg}$ |
| Elementary Charge | $e = 1.60 \times 10^{-19} \text{ C}$ |
| Coulomb's Law Constant .. | $k = 8.99 \times 10^9 \text{ N}\cdot\text{m}^2/\text{C}^2$ |
| Electron Volt | $1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$ |
| Index of Refraction of Air. | $n = 1.00$ |
| Speed of Light in Vacuum. | $c = 3.00 \times 10^8 \text{ m/s}$ |
| Planck's Constant | $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ $h = 4.14 \times 10^{-15} \text{ eV}\cdot\text{s}$ |
| Atomic Mass Unit | $u = 1.66 \times 10^{-27} \text{ kg}$ |

Physics Principles

- Uniform motion ($\vec{F}_{\text{net}} = 0$)
- Accelerated motion ($\vec{F}_{\text{net}} \neq 0$)
- Uniform circular motion (\vec{F}_{net} is radially inward)
- Work-energy theorem
- Conservation of momentum
- Conservation of energy
- Conservation of mass-energy
- Conservation of charge
- Conservation of nucleons
- Wave-particle duality

Particles

| Alpha Particle..... | Charge | Mass |
|---------------------|--------|-----------------------------------|
| Electron..... | $+2e$ | $6.65 \times 10^{-27} \text{ kg}$ |
| Proton..... | $-1e$ | $9.11 \times 10^{-31} \text{ kg}$ |
| Neutron..... | $+1e$ | $1.67 \times 10^{-27} \text{ kg}$ |
| Neutron..... | 0 | $1.67 \times 10^{-27} \text{ kg}$ |

First-Generation Fermions

| | Charge | Mass |
|--|-----------------|------------------------------|
| Electron..... | $-1e$ | $0.511 \text{ MeV}/c^2$ |
| Positron | $+1e$ | $0.511 \text{ MeV}/c^2$ |
| Electron neutrino, ν_e | 0 | $< 50 \text{ eV}/c^2$ |
| Electron antineutrino, $\bar{\nu}_e$ | 0 | $< 50 \text{ eV}/c^2$ |
| Up quark, u..... | $+\frac{2}{3}e$ | $\sim 5 \text{ MeV}/c^{2*}$ |
| Anti-up antiquark, \bar{u} | $-\frac{2}{3}e$ | $\sim 5 \text{ MeV}/c^{2*}$ |
| Down quark, d..... | $-\frac{1}{3}e$ | $\sim 10 \text{ MeV}/c^{2*}$ |
| Anti-down antiquark, \bar{d} | $+\frac{1}{3}e$ | $\sim 10 \text{ MeV}/c^{2*}$ |

*Current models seem to suggest a significantly lower mass of these quarks than those in this table.

Prefixes Used with SI Units

| Prefix | Symbol | Exponential Value |
|-------------|--------|-------------------|
| atto | a | 10^{-18} |
| femto | f | 10^{-15} |
| pico..... | p | 10^{-12} |
| nano..... | n | 10^{-9} |
| micro | μ | 10^{-6} |
| milli..... | m | 10^{-3} |
| centi..... | c | 10^{-2} |
| deci..... | d | 10^{-1} |
| deka | da | 10^1 |
| hecto | h | 10^2 |
| kilo..... | k | 10^3 |
| mega | M | 10^6 |
| giga..... | G | 10^9 |
| tera..... | T | 10^{12} |

EQUATIONS

Kinematics

$$\begin{aligned} \bar{v}_{\text{ave}} &= \frac{\Delta \bar{d}}{\Delta t} & \bar{d} &= \bar{v}t - \frac{1}{2}at^2 \\ \bar{a}_{\text{ave}} &= \frac{\Delta \bar{v}}{\Delta t} & \bar{d} &= \left(\frac{\bar{v}_1 + \bar{v}_2}{2} \right) t \\ \bar{d} &= \bar{v}t + \frac{1}{2}at^2 & v_f^2 &= v_i^2 + 2ad \\ |\bar{v}_c| &= \frac{2\pi r}{T} & |\bar{a}_c| &= \frac{v^2}{r} = \frac{4\pi^2 r}{T^2} \end{aligned}$$

Dynamics

$$\begin{aligned} \bar{\vec{F}}_{\text{net}} &= m\bar{\vec{a}} & |\bar{\vec{F}}_g| &= \frac{Gm_1m_2}{r^2} \\ |\bar{\vec{F}}_f| &= \mu|\bar{\vec{F}}_N| & |\bar{g}| &= \frac{Gm}{r^2} \\ \bar{\vec{F}}_s &= -kx & \bar{g} &= \frac{\bar{\vec{F}}_g}{m} \end{aligned}$$

Momentum and Energy

$$\begin{aligned} \bar{p} &= m\bar{v} & E_k &= \frac{1}{2}mv^2 \\ \bar{F}\Delta t &= m\Delta\bar{v} & E_p &= mgh \\ W &= |\bar{F}||\bar{d}|\cos\theta & E_p &= \frac{1}{2}kx^2 \\ W &= \Delta E \\ P &= \frac{W}{T} \end{aligned}$$

Waves

$$\begin{aligned} T &= 2\pi\sqrt{\frac{m}{k}} & m &= \frac{h_i}{h_o} = \frac{-d_i}{d_o} \\ T &= 2\pi\sqrt{\frac{l}{g}} & \frac{1}{f} &= \frac{1}{d_o} + \frac{1}{d_i} \\ T &= \frac{1}{f} & \frac{\sin\theta_1}{\sin\theta_2} &= \frac{n_2}{n_1} = \frac{v_1}{v_2} = \frac{\lambda_1}{\lambda_2} \\ v &= f\lambda & \lambda &= \frac{d \sin\theta}{n} \\ f &= \left(\frac{v}{v \pm v_s} \right) f_s & \lambda &= \frac{xd}{nl} \end{aligned}$$

Electricity and Magnetism

$$\begin{aligned} |\bar{\vec{F}}_e| &= \frac{kq_1q_2}{r^2} & \Delta V &= \frac{\Delta E}{q} \\ |\bar{\vec{E}}| &= \frac{kq}{r^2} & I &= \frac{q}{T} \\ \bar{\vec{E}} &= \frac{\bar{\vec{F}}_e}{q} & |\bar{\vec{F}}_m| &= I l |\bar{B}| \\ |\bar{\vec{E}}| &= \frac{\Delta V}{\Delta d} & |\bar{\vec{F}}_m| &= qv_{\perp} |\bar{B}| \end{aligned}$$

Atomic Physics

$$\begin{aligned} W &= hf_0 & E &= hf = \frac{hc}{\lambda} \\ E_{k,\text{max}} &= q_e V_{\text{stop}} & N &= N_0 \left(\frac{1}{2} \right)^n \end{aligned}$$

Quantum Mechanics and Nuclear Physics

$$\begin{aligned} \Delta E &= \Delta mc^2 & E &= pc \\ p &= \frac{h}{\lambda} & \Delta\lambda &= \frac{h}{mc}(1 - \cos\theta) \end{aligned}$$

Trigonometry and Geometry

$$\begin{aligned} \sin\theta &= \frac{\text{opposite}}{\text{hypotenuse}} & \text{Line} & \\ \cos\theta &= \frac{\text{adjacent}}{\text{hypotenuse}} & m &= \frac{\Delta y}{\Delta x} \\ \tan\theta &= \frac{\text{opposite}}{\text{adjacent}} & y &= mx + b \\ c^2 &= a^2 + b^2 & \text{Area} & \\ \frac{a}{\sin A} &= \frac{b}{\sin B} = \frac{c}{\sin C} & \text{Rectangle} &= lw \\ c^2 &= a^2 + b^2 - 2ab \cos C & \text{Triangle} &= \frac{1}{2}ab \\ c^2 &= a^2 + b^2 - 2ab \cos C & \text{Circle} &= \pi r^2 \\ c^2 &= a^2 + b^2 - 2ab \cos C & \text{Circumference} & \\ & & \text{Circle} &= 2\pi r \end{aligned}$$

Graphing Calculator Window Format

$$\begin{aligned} x: & [X_{\text{min}}, X_{\text{max}}, X_{\text{scl}}] \\ y: & [Y_{\text{min}}, Y_{\text{max}}, Y_{\text{scl}}] \end{aligned}$$

Periodic Table of the Elements

| | | | | | | | | | | | | | | | | | |
|-------------------------------------|------------------------------------|--|---|-------------------------------------|--------------------------------------|------------------------------------|--------------------------------------|--------------------------------------|--|---------------------------------------|-------------------------------------|---------------------------------------|--|--|---------------------------------------|--------------------------------------|---------------------------------------|
| 1 H 1.01 hydrogen | 2 He 4.00 helium | | | | | | | | | | | | | | | | |
| 3 Li 6.94 lithium | 4 Be 9.01 beryllium | 5 B 10.81 boron | 6 C 12.01 carbon | 7 N 14.01 nitrogen | 8 O 16.00 oxygen | 9 F 19.00 fluorine | 10 Ne 20.18 neon | | | | | | | | | | |
| 11 Na 22.99 sodium | 12 Mg 24.31 magnesium | 13 Al 26.98 aluminum | 14 Si 28.09 silicon | 15 P 30.97 phosphorus | 16 S 32.07 sulfur | 17 Cl 35.45 chlorine | 18 Ar 39.95 argon | | | | | | | | | | |
| 19 K 39.10 potassium | 20 Ca 40.08 calcium | 21 Sc 44.96 scandium | 22 Ti 47.87 titanium | 23 V 50.94 vanadium | 24 Cr 52.00 chromium | 25 Mn 54.94 manganese | 26 Fe 55.85 iron | 27 Co 58.93 cobalt | 28 Ni 58.69 nickel | 29 Cu 63.55 copper | 30 Zn 65.39 zinc | 31 Ga 69.72 gallium | 32 Ge 72.64 germanium | 33 As 74.92 arsenic | 34 Se 78.96 selenium | 35 Br 79.90 bromine | 36 Kr 83.80 krypton |
| 37 Rb 85.47 rubidium | 38 Sr 87.62 strontium | 39 Y 88.91 yttrium | 40 Zr 91.22 zirconium | 41 Nb 92.91 niobium | 42 Mo 95.94 molybdenum | 43 Tc (98) technetium | 44 Ru 101.07 ruthenium | 45 Rh 102.91 rhodium | 46 Pd 106.42 palladium | 47 Ag 107.87 silver | 48 Cd 112.41 cadmium | 49 In 114.82 indium | 50 Sn 118.71 tin | 51 Sb 121.75 antimony | 52 Te 127.60 tellurium | 53 I 126.90 iodine | 54 Xe 131.29 xenon |
| 55 Cs 132.91 cesium | 56 Ba 137.33 barium | 57-71 | 72 Hf 178.49 hafnium | 73 Ta 180.95 tantalum | 74 W 183.84 tungsten | 75 Re 186.21 rhenium | 76 Os 190.23 osmium | 77 Ir 192.22 iridium | 78 Pt 195.08 platinum | 79 Au 196.97 gold | 80 Hg 200.59 mercury | 81 Tl 204.38 thallium | 82 Pb 207.21 lead | 83 Bi 208.98 bismuth | 84 Po (209) polonium | 85 At (210) astatine | 86 Rn (222) radon |
| 87 Fr (223) francium | 88 Ra (226) radium | 89-103 | 104 Rf (261) rutherfordium | 105 Db (262) dubnium | 106 Sg (266) seaborgium | 107 Bh (264) bohrium | 108 Hs (277) hassium | 109 Mt (268) meitnerium | 110 Ds (271) darmstadtium | 111 Rg (272) roentgenium | 112 Uub (285) ununbium | 113 Uut (284) ununtrium | 114 Uuq (289) ununquadium | 115 Uup (288) ununpentium | 116 Uuh (292) ununhexium | 117 Uus (?) ununseptium | 118 Uuo (294) ununoctium |
| 57 La 138.91 lanthanum | 58 Ce 140.12 cerium | 59 Pr 140.91 praseodymium | 60 Nd 144.24 neodymium | 61 Pm (145) promethium | 62 Sm 150.36 samarium | 63 Eu 151.96 europium | 64 Gd 157.25 gadolinium | 65 Tb 158.93 terbium | 66 Dy 162.50 dysprosium | 67 Ho 164.93 holmium | 68 Er 167.26 erbium | 69 Tm 168.93 thulium | 70 Yb 173.04 ytterbium | 71 Lu 174.97 lutetium | 101 Md (258) mendelevium | 102 No (259) nobelium | 103 Lr (262) lawrencium |
| 89 Ac (227) actinium | 90 Th 232.04 thorium | 91 Pa 231.04 protactinium | 92 U 238.03 uranium | 93 Np (237) neptunium | 94 Pu (244) plutonium | 95 Am (243) americium | 96 Cm (247) curium | 97 Bk (247) berkelium | 98 Cf (251) californium | 99 Es (252) einsteinium | 100 Fm (257) fermium | 101 Md (258) mendelevium | 102 No (259) nobelium | 103 Lr (262) lawrencium | | | |

Key
Atomic number → 3 Li → Symbol

Atomic molar mass (g/mol) → 6.94
Name → lithium

Based on ^{12}C
() Indicates mass of the most stable isotope