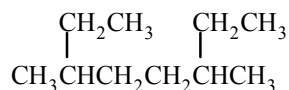
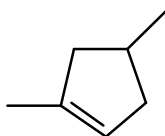


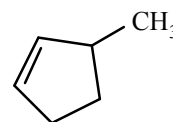
1. [15 pts] Give the IUPAC name of the following compounds.



\_\_\_\_\_



\_\_\_\_\_



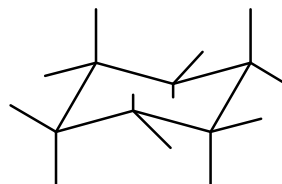
\_\_\_\_\_

2. (10 pts). How many stereoisomers are possible for 2,3-dibromobutane. Explain your answer.

3. [15 pts] Draw the structure of the following compounds.

a) *cis*-1,2-dimethylcyclopentane

b) *trans*-1,4-dimethylcyclohexane  
**(draw MOST STABLE chair form)**



c) (Z)-3-bromo-3-heptene

4. (15 pts) Write an overall equation to demonstrate the free-radical chlorination of 2,3-dimethylbutane. Include conditions and the production of the **major** organic product

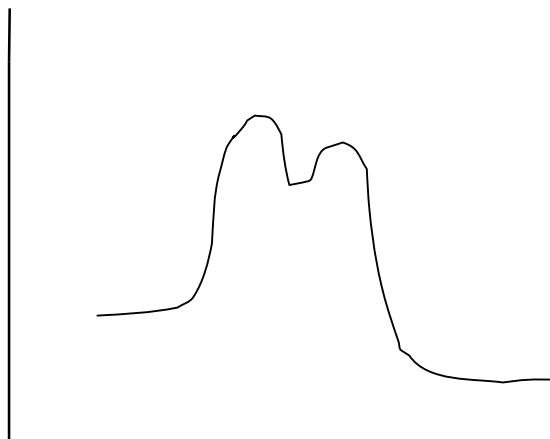
b. How many products are possible? \_\_\_\_\_

5. (10 pts) Acid HA has a  $\text{pK}_a = 20$ ; acid HB has a  $\text{pK}_a = 10$ . If NaA is added to HB, will the position of equilibrium lie to the right or left. Explain your answer.

6. [10 pts] Draw the following structures.

a) An acid and an ester, each having the molecular formula  $C_3H_6O_2$ .

7. (24 pts) Use the following diagram to answer the following: (Positions refer to I, II, III, IV, V)



What does I represent in the diagram \_\_\_\_\_

What does II represent in the diagram? \_\_\_\_\_

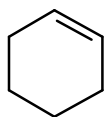
Which position(s) in the diagram represent(s) (an) intermediate(s)? \_\_\_\_\_

Which position(s) in the diagram represent(s) (a) transition state(s)? \_\_\_\_\_

Is the reaction endomermic or exothermic. \_\_\_\_\_

The difference in energy between which two points represents the energy of activation ?  
\_\_\_\_\_

For the following reaction, what are the appropriate structures at III, and V?



+ HBr

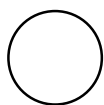
\_\_\_\_\_

III

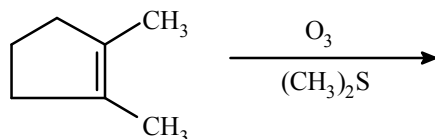
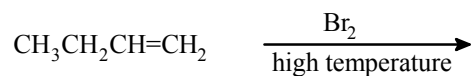
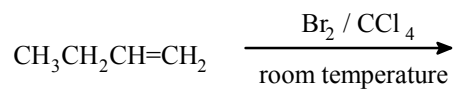
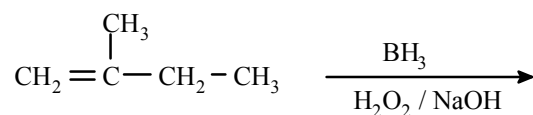
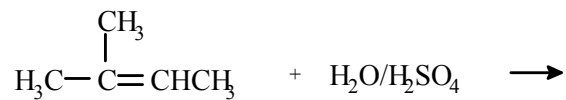
\_\_\_\_\_

V

8. (10 POINTS) Draw a Newman projection representing the **most stable** conformation of 1-bromo-2-methylpropane (as viewed along  $C_1$  to  $C_2$ )

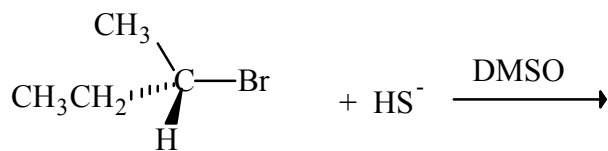


9. [25 pts] Draw the major organic product in each of the following reactions (5 points each).



10. (15 pts) For the **elimination** reaction of 2-bromo-3-methylbutane with  $\text{CH}_3\text{CH}_2\text{O}-\text{K}^+$  in  $\text{CH}_3\text{CH}_2\text{OH}$ , write an equation showing this reaction. **Include all possible products.** Circle the major product.

11. (15 pts) Give the product(s) expected in the following reaction. Indicate if the mechanism is  $\text{S}_{\text{N}}1$  or  $\text{S}_{\text{N}}2$ . **Indicate the stereochemistry for the product(s)**



12. (16 pts) Starting with **ethane**, show how you would produce an **alcohol**. you may use reagents, solvents, conditions, as necessary. You must write chemical equations showing valid reactions.