# EFFECT OF HEAT TREATMENT AND LASER SURFACE TREATMENT ON THE CORROSION BEHAVIOR OF STAINLESS STEELS

by

**Chan Weng Kin** 



## 2010/2011



Faculty of Science and Technology University of Macau

## EFFECT OF HEAT TREATMENT AND LASER SURFACE TREATMENT ON THE CORROSION BEHAVIOR OF STAINLESS STEELS

by

Chan Weng Kin

A thesis submitted in partial fulfillment of the requirements for the degree of

M. Sc. In Electromechanical Engineering

Faculty of Science and Technology University of Macau

2011

Approved by

Supervisor

Co-supervisor

Date\_\_\_\_

In presenting this thesis in partial fulfillment of the requirements for a Master's degree at the University of Macau, I agree that the Library and the Faculty of Science and Technology shall make its copies freely available for inspection. However, reproduction of this thesis for any purposes or by any means shall not be allowed without my written permission. Authorization is sought by contacting the author at

Address: PATIO PILOTO 19-23, ED. FU VA GARDEN, 4 ANDAR G, MACAU Telephone: (853) 66983661 Fax: E-mail: ma96540@umac.mo, ericchan0627@hotmail.com



### University of Macau

### Abstract

## EFFECT OF HEAT TREATMENT AND LASER SURFACE TREATMENT ON THE CORROSION BEHAVIOR OF STAINLESS STEELS

by Chan Weng Kin

Thesis Supervisor: Prof. Kwok Chi Tat Thesis Co-supervisor: Dr. Lo Kin Ho M. Sc. In Electromechanical Engineering

In the present study, the effect of heat and laser treatment on intergranular corrosion (IGC) and pitting corrosion behavior of various austenitic stainless steels ASSs (UNS S30400, S31603, S32100, S34700, FeCrMn), duplex stainless steels DSSs (S31803 and S32950) and super duplex stainless steel SDSS (S32760) were investigated. Double loop electrochemical potentiokinetic reactivation (DL-EPR) test was performed by using a potentiostat on aged ASSs in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 0.01 M KSCN solution at 25 °C whereas for aged DSSs and SDSS, a more aggressive solution (2 M H<sub>2</sub>SO<sub>4</sub>, 0.01 M KSCN and 0.5 M NaCl at 25 °C) was used. SEM examination revealed that intergranular attack occurs at the Cr-depleted grain boundaries for the aged ASSs while attack takes place at the phase boundaries of the intermetallic  $\sigma$  phase and metallic phase ( $\gamma$ ,  $\delta$ ,  $\gamma_2$ ) for the aged DSSs. Cyclic potentiodynamic polarization was also carried out for investigating the pitting corrosion behavior of various aged stainless steels in 3.5 wt% NaCl solution at 25°C. For FeCrMn, increase in the aging time significantly deteriorates the pitting resistance and the specimens

even do not passive when the aging time is longer than 50 h. For aged S31803 and S32950, the pitting resistance is also deteriorated as evidenced by the decrease in pitting potential. Similar results are obtained for aged S32760, but higher repassivation ability was observed mainly due to the present of more passivable alloying elements.

Laser surface melting (LSM) was done on various stainless steels by a 2.5-kW CW Nd:YAG laser and a 2.3-kW diode laser. Different laser processing parameters were applied to melt the surface of FeCrMn and S32950 for optimizing the LSM conditions for corrosion resistance. The pitting resistance of the aged S32950 after LSM with a higher scanning speed and lower power of the laser is higher than that of the ones fabricated with a lower processing speed or higher power. In addition, the aged ASSs were essentially austenitic with some  $\delta$  (except FeCrMn) but the chromium carbides were completely removed after LSM. For the aged DSS and SDSS after LSM,  $\delta$  became the major phase and the  $\delta/\gamma$  phase balance was disturbed but the  $\sigma$  and  $\gamma_2$  phases were eliminated. After LSM, the IGC resistance of the aged ASSs was found to be considerably improved as reflected by the reduction in degree of sensitization (DOS). This could be attributed to a more homogenous microstructure and the redissolution of chromium carbides. The IGC resistance of aged DSSs and SDSS was also significantly enhanced as indicated by the decrease in DOS. This could be attributed to a more homogenous microstructure and to the redissolution of  $\sigma$ and  $\gamma_2$  phases. Furthermore, the effects of re-aging after LSM were also investigated on FeCrMn and S32950. Both of them showed higher pitting resistance as compared with that of the aged specimens.

## TABLE OF CONTENTS

| LIST OF FIGURESvi   |
|---|
| LIST OF TABLES  |
| LIST OF ABBREVIATIONSxv   |
| LIST OF PUBLICATIONSxvi   |
| ACKNOWLEDGEMENTSxviii   |
| CHAPTER 1: Introduction 1.1 Research Background1 1.2 Objectives |
| CHAPTER 2: Literature Review                                    |
| 2.1 Role of alloying elements in stainless steels4              |
| 2.1.1 Ferritic stainless steels                                 |
| 2.1.2 Austenitic stainless steels                               |
| 2.1.2.1 Common grade ASSs7                                      |
| 2.1.2.2 Stabilized grade ASSs                                   |
| 2.1.2.3 Ni-free grade ASSs                                      |
| 2.1.3 Duplex stainless steels                                   |
| 2.1.3.1 Lean grade DSSs11                                       |
| 2.1.3.2 Super DSSs and hyper DSSs                               |
| 2.2 Effect of thermal aging on stainless steels                 |
| 2.2.1 Sensitization of austenitic stainless steels              |
| 2.2.2 Sensitization of duplex stainless steels                  |

| 2.3 Laser surfa | ace treatment                                     | 17 |
|-----------------|---|----|
| 2.3.1 Las       | er surface melting                                |    |
| 2.3.2 Adv       | vantages of LSM                                   | 19 |
| 2.3.3 Hea       | t transfer model of LSM                           | 20 |
| 2.3.4 App       | lications of LSM                                  | 23 |
| 2.4 Corrosion   | of stainless steels                               | 24 |
| 2.4.1 Ger       | neral corrosion                                   | 25 |
| 2.4.2 Pitt      | ing corrosion                                     | 27 |
| 2.4.2           | 2.1 Pitting corrosion test                        | 27 |
| 2.4.3 Inte      | rgranular corrosion                               | 30 |
| 2.4.3           | 3.1 Intergranular corrosion test                  | 31 |
| 2.5 Review of   | current research of LSM of stainless steels       |    |
| 2.5.1 IGC       | behaviors of LSM of stainless steels              |    |
| 2.5.1           | 1.1 Effect of LSM, LSA and LC                     | 34 |
| 2.5.1           | 1.2 LSM of weldments                              | 35 |
| 2.5.1           | 1.3 Effect of cooling rate (δ-ferrite content)    | 35 |
| 2.5.1           | 1.4 Effect of cold work                           | 36 |
| 2.5.1           | 1.5 Effect of processing parameters               | 37 |
| 2.5.1           | 1.6 Effect of resensitization and GBE study       | 37 |
| 2.5.2 Pitt      | ing corrosion behavior of LSM of stainless steels |    |
| 2.5.2           | 2.1 Effect of heat treatment after LSM            |    |
| 2.5.2           | 2.2 Effect of processing parameters               |    |
| 2.5.2           | 2.3 Effect of shielding gas                       | 40 |
| CHAPTER 3: Expe | erimental Details                                 | 46 |
| 3.1 Materials a | and isothermal aging                              | 46 |

| 3.2 Laser surface melting  | 47          |
|--|-------------|
| 3.3 Corrosion test   | 49          |
| 3.3.1 Pitting corrosion test   | 50          |
| 3.3.2 DL-EPR test  | 51          |
| 3.4 Microstructural analysis   |             |
| 3.5 Hardness measurement   | 53          |
| CHAPTER 4: Results and Discussion: Effect of Heat Treatment on Intergra  | anular and  |
| Pitting Corrosion Behavior of Stainless Steels                           | 55          |
| 4.1 Effect of isothermal aging on ASS, DSS and SDSS                      | 55          |
| 4.1.1 Metallographic and microstructural analysis                        | 55          |
| 4.1.2 Intergranular corrosion behavior                                   | 68          |
| 4.1.2.1 IGC of S34700 for aging time up to 720 h                         | 68          |
| 4.1.2.2 IGC of various ASSs and DSSs aged for 40 h                       | 73          |
| 4.1.3 Pitting corrosion behavior   | 80          |
| 4.1.3.1 Aged ASS S34700  |             |
| 4.1.3.2 Aged ASS FeCrMn  |             |
| 4.1.3.3 Aged DSS S31803  | 90          |
| 4.1.3.4 Aged DSS S32950  | 95          |
| 4.1.3.5 Aged SDSS S32760   | 99          |
| CHAPTER 5: Effect of Laser Surface Melting on Intergranular and Pitting  | Corrosion   |
| Behavior of Stainless Steels   | 105         |
| 5.1 Effect of laser processing parameters on the pitting corrosion behav | vior of ASS |
| FeCrMn and DSS S32950  | 105         |
| 5.1.1 Metallographic and microstructural analysis of FeCrMn              | 105         |
| 5.1.1.1 Single track LSM   |             |

| 5.1.1.2 LSM of full surface  | 110   |
|--|-------|
| 5.1.1.3 Metallographic and microstructural analysis                        | 112   |
| 5.1.2 Pitting corrosion behavior of FeCrMn                                 | 117   |
| 5.1.3 Metallographic and microstructural analysis of S32950                | 125   |
| 5.1.4 Pitting corrosion behavior of S32950                                 | 131   |
| 5.1.4.1 Aged S32950 after LSM  | 131   |
| 5.1.4.2 Aged S32950 for 200 h after LSM                                    | 136   |
| 5.2 IGC behavior of aged ASSs and DSSs after LSM                           | 140   |
| 5.2.1 Metallographic and microstructural analysis                          | 140   |
| 5.2.2 IGC behavior   | 151   |
| 5.2.2.1 Aged ASSs after LSM  | 157   |
| 5.2.2.2 Aged DSSs and SDSS after LSM                                       | 160   |
| 5.3 Effect of re-aging on the pitting corrosion behavior of ASS FeCrMn and | d DSS |
| S32950   | 163   |
| 5.3.1 Effect of re-aging on the pitting corrosion behavior of ASS          |       |
| FeCrMn   | 163   |
| 5.3.2 Effect of re-aging on the pitting corrosion behavior of DSS          |       |
| \$32950  | 167   |
| CHAPTER 6: Conclusions   | 172   |
| CHAPTER 7: Suggestions for further works                                   | 175   |
| REFERENCES   | 176   |
| Chapter 1  | 176   |
| Chapter 2  | 179   |
| Chapter 3  | 187   |
| Chapter 4  | 188   |

|     | Chapter 5.1 |  |
|-----|-------------|--|
|     | Chapter 5.2 |  |
|     | Chapter 5.3 |  |
| API | PENDIX      |  |



### LIST OF FIGURES

| Figure 2.1 An optical micrograph showing a sensitized ASS14                         |
|---|
| Figure 2.2 Isothermal precipitation diagram for DSS 2205, annealed at 1050 °C       |
| (Duplex grades 2304 and 2507 are shown for comparison)                              |
| Figure 2.3 Classification of laser remanufacturing technology                       |
| Figure 2.4 LSM processing diagram   |
| Figure 2.5 Figure 2.5 Schematic illustration of flat free surface of LSM21          |
| Figure 2.6 A corrosion pit on the surface of a UNS S32304 stainless steel27         |
| Figure 2.7 Representative cyclic potentiodynamic polarization curves for UNS        |
| S30400 in 3.56 wt% NaCl solution at 25 ± 1 °C29                                     |
| Figure 2.8 Hypothetical cathodic and anodic polarization plot for                   |
| determininglocalized corrosion parameters   |
| Figure 2.9 A Typical IGC morphology of S30400 after DL-EPR test32                   |
| Figure 2.10 Principle of DL-EPR technique and IGC sensitization criterion: (a)      |
| non-sensitized material and (b) sensitized material                                 |
| Figure 3.1 Focus point illustration   |
| Figure 3.2 Graphical illustration of calculating degree of overlapping49            |
| Figure 3.3 Pitting corrosion test criteria for the stainless steels                 |
| Figure 3.4 DL-EPR test criteria for the stainless steels                            |
| Figure 4.1.1 SEM micrographs of S34700 aged at 600 °C for 720 h (etched by oxalic   |
| acid at 5V for 20 s)  |
| Figure 4.1.2 SEM micrographs of FeCrMn aged at 600 °C for (a) 4 h, (b) 50 h and (c) |
| 200 h   |

| Figure 4.1.3 SEM micrographs of S31803 aged at 800 °C for (a) solution-annealed, (b)  |
|---|
| 1 h, (c) 4 h, (d) 50 h, (e) 100 h and (f) 200 h (for (b) to (e), BSE image with EDX   |
| results in wt%), at a magnification of 500X60   |
| Figure 4.1.4 BSE micrographs of S31803 aged at 800 °C for (a) solution-annealed, (b)  |
| 1 h, (c) 4 h, (d) 50 h, (e) 100 h and (f) 200 h, at a magnification of 1000X61        |
| Figure 4.1.5 SEM micrographs of S32950 for (a) annealed (b) 4 h, (c) 50 h and (d)     |
| 200 h aged at 800°C62   |
| Figure 4.1.6 XRD spectra of various aging time of S3295063                            |
| Figure 4.1.7 SEM micrographs of S32760 aged at 800°C for (a) solution-annealed, (b)   |
| 1 h, (c) 4 h, (d) 50 h, (e) 100 h and (f) 200 h (For (b) to (e), BSE image with EDX   |
| results in wt%), at a magnification of 500X67   |
| Figure 4.1.8 BSE micrographs of S32760 aged at 800 °C for (a) solution-annealed, (b)  |
| 1 h, (c) 4 h, (d) 50 h, (e) 100 h and (f) 200 h, at a magnification of 1000X68        |
| Figure 4.1.9 DL-EPR curves for aged S34700 up to 720 h in 0.5 M $H_2SO_4$ + 0.01 M    |
| KSCN solution at 25 °C69  |
| Figure 4.1.10 DOS value plot against aging time of \$34700                            |
| Figure 4.1.11 Corrosion morphologies of S34700 after DL-EPR test which aged at (a)    |
| 40 h, (b) 200 h, (c) 333 h, (d) 455 h, (e) 575 h and (f) 720 h71                      |
| Figure 4.1.12 (a) Corrosion morphology of aged S34700 after DL-EPR test and (b)       |
| EDX spectrum of Nb-rich phase72   |
| Figure 4.1.13 Schematic illustration of newly proposed IGC mechanism developed by     |
| Cr-depletion due to segregation of the un-reacted Cr atoms in the grain boundary area |
|   |
| Figure 4.1.14 DL-EPR curves for (a) aged S30400, S31603, S32100, S34700 (aged         |
| for 720 h) and FeCrMn before LSM in 0.5 M $H_2SO_4$ + 0.01 M KSCN solution at 25      |

| $^{\circ}\text{C}$ and (b) aged S31803, S32950 and S32760 before LSM in 2 M $\text{H}_{2}\text{SO}_{4}$ + 0.01 M |
|--|
| KSCN + 0.5M NaCl solution at 25 °C75   |
| Figure 4.1.15 Corrosion morphologies of aged ASSs after DL-EPR test (a) S30400, (b)                              |
| S31603, (c) S32100, (d) S34700, and (e) FeCrMn78   |
| Figure 4.1.16 Corrosion morphologies of aged DSSs and SDSS after DL-EPR test (a)                                 |
| \$31803, (b) \$32950 and (c) \$3276080   |
| Figure 4.1.17 Cyclic polarization curves of (a) S34700, (b) FeCrMn, (c) S31803 (d)                               |
| S32950 and (d) S32760 aged at various time up to 200 h   |
| Figure 4.1.18 Corrosion parameters for aged S34700 up to 200 h (a) $E_{corr}$ , (b) $I_{corr}$ , (c)             |
| I <sub>pass</sub> (d) E <sub>pit</sub> and (e) E <sub>prot</sub>   |
| Figure 4.1.19 Corrosion morphology of aged S34700 for (a) annealed (b) 4 h, (c) 50 h                             |
| and (d) 200 h after cyclic polarization test   |
| Figure 4.1.20 Corrosion parameters for aged FeCrMn up to 200 h (a) $E_{corr}$ , (b) $I_{corr}$ , (c)             |
| I <sub>pass</sub> (d) E <sub>pit</sub> and (e) E <sub>prot</sub>   |
| Figure 4.1.21 Corrosion morphology of Aged FeCrMn for (a) annealed (b) 4h, (c) 50h                               |
| and (d) 200 h after cyclic polarization test90   |
| Figure 4.1.22 Corrosion parameters for aged S31803 aged up to 200h (a) $E_{corr}$ , (b) $I_{corr}$ ,             |
| (c) I <sub>pass</sub> (d) E <sub>pit</sub> and (e) E <sub>prot</sub> .   |
| Figure 4.1.23 Corrosion morphology of aged S31803 for (a) annealed (b) 4 h, (c) 50 h                             |
| and (d) 200 h after cyclic polarization test at a magnification of 200X94  |
| Figure 4.1.24 Corrosion morphology of aged S31803 for (a) annealed (b) 4 h, (c) 50 h                             |
| and (d) 200 h after cyclic polarization test at a magnification of 1000X95                                       |
| Figure 4.1.25 Corrosion parameters for aged S32950 aged up to 200 h (a) $E_{corr}$ , (b)                         |
| I <sub>corr</sub> , (c) I <sub>pass</sub> (d) E <sub>pit</sub> and (e) E <sub>prot</sub> 98                      |
| Figure 4.1.26 Corrosion morphology of aged S32950 for (a) 4 h, (b) 50 h and (c) 200                              |

| h after cyclic polarization test98  |
|---|
| Figure 4.1.27Corrosion parameters for S32760 aged up to 200 h (a) $E_{corr}$ , (b) $I_{corr}$ , (c) |
| $I_{pass}$ , (d) $E_{pit}$ and (e) $E_{prot}$ 102   |
| Figure 4.1.28 Corrosion morphology of aged S32760 for (a) annealed (b) 4 h, (c) 50 h                |
| and (d) 200 h after cyclic polarization test at a magnification of 200X103                          |
| Figure 4.1.29 Corrosion morphology of aged S32760 for (a) annealed (b) 4h, (c) 50h                  |
| and (d) 200h after cyclic polarization test at a magnification of 1000X104                          |
| Figure 5.1.1 Cross-sectional views of various LSM specimens107                                      |
| Figure 5.1.2 Data plots of various sets of laser processing parameters                              |
| Figure 5.1.3 Surface appearances and cross-sectional views of the specimens                         |
| fabricated at various laser processing conditions after LSM112                                      |
| Figure 5.1.4 EDX spectrum of as-received FeCrMn stainless steel,                                    |
| Figure 5.1.5 A typical transverse cross-sectional view of LSM specimen (L20)114                     |
| Figure 5.1.6 XRD spectrum of LSM specimen (L20)   |
| Figure 5.1.7 Microstructures of melt zone of laser-melted FeCrMn for the processing                 |
| parameter (a) L17 (b) L20, (c) L25 and (d) L28116   |
| Figure 5.1.8 EDX spectra of (a) melt pool and (b) substrate for L17116                              |
| Figure 5.1.9 Hardness profile along the depth of the cross-section of melt pool (a) L17             |
| and (b) L28117  |
| Figure 5.1.10 Cyclic polarization curves of laser-melted FeCrMn with various laser                  |
| processing parameters119  |
| Figure 5.1.11 Corrosion parameters of FeCrMn at various scanning speeds                             |
| Figure 5.1.12 Corrosion parameters of FeCrMn at various beam diameters124                           |
| Figure 5.1.13 Corrosion morphology of (a) as-received and (b) L20 after cyclic                      |
| polarization test   |

| Figure 5.1.14 SEM micrographs of aged S32950 (for 200 h) laser-melted with   |
|--|
| processing condition L1 (a) cross-sectional view; and (b) microstructure of melt zone.   |
|  |
| Figure 5.1.15 SEM micrographs of aged S32950 (for 200h) laser-melted with  |
| processing condition L2 (a) cross-sectional view; and (b) microstructure of melt zone.   |
|  |
| Figure 5.1.16 SEM micrographs of aged S32950 (for 200 h) laser-melted with   |
| processing condition L3 (a) cross-sectional view; and (b) microstructure of melt zone.   |
|  |
| Figure 5.1.17 XRD spectra of various aging time after LSM of parameter L1128   |
| Figure 5.1.18 XRD spectra aged at 200 h after LSM with various laser parameters.129  |
| Figure 5.1.19 Harness profiles of S32950 aged for 200 h after LSM with L1, L2 and  |
| L3   |
| Figure 5.1.20 Polarization curves of aged S32950 laser-melted with L1132   |
| Figure 5.1.21 Polarization curves of aged S32950 laser-melted with L2133   |
| Figure 5.1.22 Polarization curves of aged S32950 laser-melted with L3134   |
| Figure 5.1.23 Corrosion parameters for aged S32950 after LSM (a) $E_{corr}$ (b) $E_{pit}$ and (c)  |
| E <sub>prot</sub>  |
| Figure 5.1.24 Corrosion parameters for aged S32950 after LSM (a) $I_{corr}$ and (b) $I_{pass}.$  |
|  |
| Figure 5.1.25 Polarization curves of laser-melted, aged and annealed S32950137   |
|  |
| Figure 5.1.26 Corrosion morphology of (a) aged S32950 for 200 h, (b) aged S32950   |
| Figure 5.1.26 Corrosion morphology of (a) aged S32950 for 200 h, (b) aged S32950 laser-melted with L1 (c) aged S32950 laser-melted with L2 (d) aged S32950 |

| Figure 5.2.1 SEM micrographs of LM-30400-S (a) cross-sectional view (OM) and (b) |
|--|
| microstructure of melt pool (SE image with EDX results)140                       |
| Figure 5.2.2 SEM micrographs of LM-31603-S (a) cross-sectional view (OM) and (b) |
| microstructure of melt pool (SE image with EDX results)141                       |
| Figure 5.2.3 SEM micrographs of LM-32100-S (a) cross-sectional view (OM) and (b) |
| microstructure of melt pool (SE image with EDX results)141                       |

Figure 5.2.4 SEM micrographs of LM-34700-S (a) cross-sectional view (OM) and (b) microstructure of melt pool (SE image with EDX results)......142 Figure 5.2.5 SEM micrographs of LM-FeCrMn-S (a) cross-sectional view (OM) and (b) microstructure of melt pool (SE image with EDX results in wt%, IB = interdendritic boundary)......142 Figure 5.2.6 SEM micrographs of LM-S31803 (a) cross-sectional view (OM), (b) microstructure of melt pool (SE image with EDX results in wt%), and (c) microstructure of sensitized zone (BSE image with EDX results in wt%)......143 Figure 5.2.7 SEM micrographs of LM-S32950-S (a) cross-sectional view (OM), (b) microstructure of melt pool (BSE image with EDX results in wt%), and (c) microstructure of aged substrate (BSE image with EDX results in wt%). .....144 Figure 5.2.8 SEM micrographs of LM-S32760-S (a) cross-sectional view (OM), (b) microstructure of melt pool (BSE image with EDX results in wt%), and (c) microstructure of aged substrate (BSE image with EDX results in wt%). ......145 Figure 5.2.9 Hardness of various stainless steels in annealed, aged and laser-melted Figure 5.2.10 DL-EPR curves for aged (a) S30400, (b) S31603, (c) S32100, (d) S34700, and (e) FeCrMn before and after LSM in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 0.01 M KSCN

| solution at 25 °C   |
|---|
| Figure 5.2 11 DL-EPR curves for aged (a) S31803, (b) S32950 and (c) S32760 before     |
| and after LSM in 2 M $H_2SO_4$ + 0.01 M KSCN + 0.5M NaCl solution at 25 °C 155        |
| Figure 5.2 12 Corrosion morphologies of aged ASSs after DL-EPR test (a) S30400, (b)   |
| S31603, (c) S32100, (d) S34700, and (e) FeCrMn (i) before LSM and (ii) after LSM.     |
|   |
| Figure 5.2.13 Corrosion morphologies of aged DSSs and SDSS after DL-EPR test (a)      |
| S31803, (b) S32950 and (c) S32760 (i) before LSM and (ii) after LSM157                |
| Figure 5.3.1 Polarization curves of laser-melted FeCrMn after re-aging for 50 h and   |
| annealed specimens  |
| Figure 5.3.2 Polarization curves of (a) L20 and re-aged L20, (b) L28 and re-aged L28. |
|   |
| Figure 5.3.3 Corrosion morphology of re-aged L20 after polarization test167           |
| Figure 5.3.4 Polarization curves of laser-melted S32950 after re-aging for 200h,      |
| annealed and aged specimens   |
| Figure 5.3.5 Polarization curves of (a) L1 and re-aged L1, (b) L2 and re-aged L2, and |
| (c) L3 and re-aged L3   |
| Figure 5.3.6 Corrosion morphology of (a) aged S32950 for 200h, (b) re-aged S32950     |
| laser-melted with L1 (c) re-aged S32950 laser-melted with L2 and (d) re-aged S32950   |
| laser-melted with L3  |

### LIST OF TABLES

| Table 2.1 A summary of research on laser surface melting of stainless steels for      |
|---|
| mitigating IGC43  |
| Table 2.2 A summary of research on LSM of stainless steels for improving pitting      |
| corrosion resistance45  |
| Table 3.1 Compositions of various stainless steels in wt%.                            |
| Table 3.2 Processing parameters for LSM of 7MoPlus47                                  |
| Table 3.3 Processing parameters for LSM of FeCrMn.    49                              |
| Table 4.1.1 DOS of S34700 at various aging time after DL-EPR test69                   |
| Table 4.1.2 DOS of various stainless steels after DL-EPR test.    74                  |
| Table 4.1.3 Corrosion parameters of S34700 from cyclic polarization test              |
| Table 4.1.4 Corrosion parameters of FeCrMn from cyclic polarization test.       89    |
| Table 4.1.5 Corrosion parameters of S31803 from cyclic polarization test              |
| Table 4.1.6 Corrosion parameters of \$32950 from cyclic polarization test             |
| Table 4.1.7 Corrosion parameters of \$32760 from cyclic polarization test             |
| Table 5.1.1 Various single-track laser processing parameters for FeCrMn.       106    |
| Table 5.1.2 Laser processing parameters and degree of overlapping for various         |
| specimens111  |
| Table 5.1.3 Chemical compositions of ASSs S30400 and FeCrMn (the content of the       |
| elements is in wt-%)114   |
| Table 5.1. 4 Chemical compositions of L17 (a) melt pool and (b) substrate116          |
| Table 5.1.5 Corrosion parameters of laser-melted FeCrMn at different laser processing |
| parameters120   |

| Table 5.1.6 Processing parameters for LSM.  | 126  |
|---|------|
| Table 5.1.7 Corrosion parameters of laser-melted S32760 (L1).   | 133  |
| Table 5.1.8 Corrosion parameters of laser-melted S32760 (L2).   | 133  |
| Table 5.1.9 Corrosion parameters of laser-melted S32760 (L3).   | 134  |
| Table 5.1.10 Corrosion parameters of laser-melted, aged and annealed S32950                             | 137  |
| Table 5.2.1 $Cr_{eq}/Ni_{eq}$ ratio, phase present and volume fraction of ferrite ( $C_{\delta}$ ) in a | ıged |
| stainless steels before and after LSM   | 150  |
| Table 5.2.2 DOS of various stainless steels after DL-EPR test.  | 152  |
| Table 5.3.1 Corrosion parameters of re-aged FeCrMn after LSM, aged for 50 h                             | and  |
| annealed samples.   | 165  |
| Table 5.3.2 Corrosion parameters of laser-melted S32950 after re-aging, annealed                        | and  |
| aged \$32950  | 168  |
|   |      |



## LIST of Abbreviations

| ASS               | Austenitic Stainless Steel                    |
|-------------------|---|
| ASTM              | America Society of Testing and Materials      |
| DOS               | Degree of Sensitization                       |
| DSS               | Duplex Stainless Steel                        |
| E <sub>corr</sub> | Corrosion Potential                           |
| EDX               | Energy Dispersive Spectroscopy                |
| E <sub>pit</sub>  | Pitting Potential DADE                        |
| E <sub>prot</sub> | Protection Potential                          |
| FSS               | Ferrite Stainless Steel                       |
| Icorr             | Corrosion Current                             |
| IGC               | Intergranular Corrosion                       |
| I <sub>pass</sub> | Passivation Current                           |
| ISO               | International Organization for Standarization |
| LC                | Laser Cladding                                |
| LSA               | Laser Surface Alloying                        |
| LSM               | Laser Surface Melting                         |
| ОСР               | Open Circuit Potential                        |
| ОМ                | Optical Microscope                            |
| РС                | Pitting Corrosion                             |
| SDSS              | Super Duplex Stainless Steel                  |
| SEM               | Scanning Electron Microscope                  |
| XRD               | X-Ray Diffractometry                          |

#### LIST OF PUBLICATIONS

### **Book Chapter**

 <u>W.K. Chan</u>, C.T. Kwok, K.H. Lo, Chapter 3. Laser surface melting of stainless steels for mitigating intergranular corrosion, in 'Laser Surface Modification of Alloys for Erosion and Corrosion Resistance', C.T. Kwok (Editor), Woodhead Publishing, ISBN 13: 9780857090157 ISBN 10: 0857090151 (to be published in 2012)

### **Journal Paper**

 C.T. Kwok, K.H. Lo, <u>W.K. Chan</u>, F.T. Cheng, H.C. Man, Effect of laser surface melting on intergranular corrosion behaviour of aged austenitic and duplex stainless steels, Corrosion Science, 53 (2011) pp.1581-1591. (SCI, Impact factor: 3.261)

### **Conference Papers**

- <u>W.K. Chan</u>, C.T. Kwok, K.H. Lo, 'Effect of Laser Surface Melting on Corrosion Behavior of Aged Duplex Stainless Steel', Proceedings of the 29th International Congress on Applications of Lasers & Electro-Optics 2010 (ICALEO '10), Anaheim, CA, USA, 27-30 Sept. (2010). (EI)
- C.T. Kwok, P.K. Wong, <u>W. K. Chan</u>, F.T. Cheng, H.C. Man, 'Laser Surface Alloying of Mn-Ni-Al Bronze for Cavitation Erosion Resistance', Proceedings of

the 29th International Congress on Applications of Lasers & Electro-Optics 2010 (ICALEO '10), Anaheim, CA, USA, 27-30 Sept. (2010). (EI)

- <u>W.K. Chan,</u> C.T. Kwok, K.H. Lo, Z. Cheng, 'Desensitization of Austenitic and Duplex Stainless Steels by Laser Surface Melting', Proceedings of the 3rd Pacific International Conference on Application of Lasers and Optics (PICALO 2010), March 23 – 25, Wuhan, China. (EI)
- P.K Wong, <u>W.K. Chan</u>, C.T. Kwok, K.H. Lo, 'Applications of laser remanufacturing for wear and corrosion resistance', Proceedings of 6th International Symposium on Environmentally Conscious Design and Inverse Manufacturing (EcoDesign 2009), December 7-9, 2009, Sapporo, Japan, pp.303-307.
- <u>W.K. Chan</u>, C.T. Kwok, K.H. Lo, S.F. Wong, 'The Critical Part of PEMFC Super Duplex Stainless Steel Bipolar Plates' 6th International Symposium on Environmentally Conscious Design and Inverse Manufacturing (EcoDesign 2009), December 7-9, 2009, Sapporo, Japan, pp.1185-1190

-342

即日

-

### ACKNOWLEDGMENTS

Firstly, I would like to express my sincere appreciation to my supervisor Prof. Kwok Chi Tat and co-supervisor Dr. Lo Kin Ho for their kind supports and guidance on my research work during graduate study. Besides conducting research work in UM, I have chance to attend two international conferences during my master study, I would like to acknowledge Prof. Kwok once again for giving me such good opportunities to open my mind and share his experience with me.

Also, I would like to thank Ms. Wong Po Kee and Mr. Zhang Bo Kai for their help and technical supports in the Corrosion and Metallography Laboratory, Laser Laboratory and SEM Laboratory; in addition, I wish to acknowledge the supports from the infrastructure of the University of Macau, the academic and administrative staff in the Department of Electromechanical Engineering, and also the financial support from Science and Technology Development Fund of Macau (FDCT).

Finally, I would like to express my sincere thanks to my beloved friends and family for their support and encouragement. Last but not least, I would like to thank God for being the guidance of my life.