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## Elements and the Periodic Table • Chapter Test B

## Elements and the Periodic Table Multiple Choice <br> Write the letter of the correct answer on the line at the left.

$\qquad$ 1. Which particles are found in the nucleus of an atom?
a. electrons and protons
b. neutrons and protons
c. neutrons and electrons
$\qquad$ 2. The number of protons in an atom is the element's
a. chemical symbol.
b. group.
c. atomic number.
$\qquad$ 3. Dmitri Mendeleev created the first
a. chemical reaction.
b. periodic table.
c. semiconductor.
$\qquad$ 4. A row of elements in the periodic table is called a
a. period.
b. group.
c. family.
$\qquad$ 5. A one or two letter representation of an element is $\mathrm{a}(\mathrm{an})$
a. chemical symbol.
b. atomic number.
c. energy level.
6. If an object can be pulled into a long wire, then the object is
a. ductile.
b. reactive.
c. easy to break.
$\qquad$ 7. When metal atoms react with the atoms of other elements, the metal atoms
a. gain neutrons.
b. share protons.
c. lose electrons.
8. Scientists make elements heavier than uranium in machines called
a. semiconductors.
b. particle accelerators.
c. lasers.
$\qquad$
$\qquad$

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$\qquad$ 9. Which is a property of nonmetals?
a. good conductor
b. malleable
c. not shiny
10. Elements that have some properties of both metals and nonmetals are called
a. semimetals.
b. noble gases.
c. alkali metals.

## Completion

Read each word in the box. In each sentence below, fill in the correct word or words. Not all of the words will be used.

| alkali metals | atomic mass | halogens |
| :--- | :--- | :--- |
| isotopes | radioactivity | noble gases |

11. Atoms with the same number of protons but different numbers of neutrons are called $\qquad$ .
12. The average mass of all the isotopes of an element is called the element's $\qquad$ .
13. The most reactive group of metals in the periodic table is the
14. The most reactive group of nonmetals whose atoms gain or share one electron in chemical reactions are the
$\qquad$ .
15. The ability of a substance to spontaneously give off radiation is called
$\qquad$ -.

## True or False

If a statement is true, write true. If it is false, write false.
$\qquad$ 16. In an atom, the number of protons is equal to the number of electrons.
$\qquad$ 17. It is impossible to predict the properties of any element by its position in the periodic table.
$\qquad$
$\qquad$
$\qquad$

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$\qquad$ 18. Heat and electric current can easily flow through metals.
19. Nonmetals can be hammered into thin sheets.
20. Radioactive tracers can be used to follow the steps of a chemical reaction.

## Using Science Skills

Use the diagram below to answer questions 21, 22, and 23.

21. Inferring The part of an atom labeled B is a(an)
a. proton.
b. neutron.
c. electron.
$\qquad$ 22. Applying Concepts Which part of an atom has the least mass?
a. A
b. B
c. C
23. Applying Concepts Which parts of an atom make up nearly all the mass of an atom?
a. $\mathrm{A}+\mathrm{C}$
b. $A+B$
c. $\mathrm{B}+\mathrm{C}$
$\qquad$
$\qquad$

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## Using Science Skills

Use the diagram below to answer questions 24 and 25.
15

| 7 <br> Nitrogen |
| :---: |
| 15 <br> Phosphorus |
|  |
| 51 <br> Antimony |
|  |

24. Inferring The chemical symbol for phosphorus is
a. 15 .
b. P.
c. Sb .
25. Applying Concepts The elements shown in the diagram are
a. part of the same group.
b. part of the same period.
c. not related.
