Side 1	Side 7 (Isotopes)
Element's Name	How many does this element have?
Element's Pronunciation	List all of them with Chemical Symbol-Mass #
Side 2 (Element as it appears on Periodic Table)	List or highlight the radioactive isotopes
Atomic Number	Side 8 (Chemical Properties)
Chemical Symbol	Elements it commonly reacts with
Atomic Mass	How does it occur naturally? (Alone or combined with what elements)
Side 3 (Appearance)	
Colored picture or drawing of element at room temperature	Side 9 (Purpose)
	How is the element use today?
Side 4 (Appearance Continued)	Side 10 (History)
Written description of physical appearance at room temperature	Year element was discovered
	Who discovered this element
Side 5 (Physical Properties)	Where the element was discovered
Density	Meaning of element's name
Melting Point	
Boiling Point	Side 11 (lon)
Specific Heat Capacity	Does it become an ion?
Control (Miles on the combet formed and the Device tie Telele)	If so, will it be positive or negative?
Side 6 (Where it can be found on the Periodic Table)	If so, will it gain or lose electrons? How many?
Small copy of the Periodic Table with its group and period highlighted or shaded and labeled.	Side 12 (Atomic Diagram)
and ponou ingringriou or oriadou and laborou.	Nucleus labeled with the number of protons and the number of neutrons of the most common atom Include all energy levels in the electron cloud

Name: ______ Date: _____ Bcience: _____ #: _____

Other information:

Each side is worth 8 points. To receive all points for each side, you must include all information for your element and it must be accurate.

Diagram the valence electrons

The dodecahedron template and a small Periodic Table can be downloaded from onmycalendar.com.

Attach an index card that includes your name (first and last), the date turned in, and your class period. On the back of the index card, include all sources used to get your information in MLA format.

Due Date: Friday, October 11, 2013