Type: Single Date:\_\_\_\_\_

Objective: Heat, Temperature and Linear Expansion of Solids

Homework: READ 12.1 - 12.4

Do CONCEPT Q. # (5, 7)

Do PROBLEMS (11, 13, 19) Ch. 12 (12/19)

How Heat Effects the Expansion and Contraction of matter

The Kelvin Temperature and Pressure – Absolute Zero

The Super Fridge

Date:

In the United States, temperatures are often expressed in degrees Fahrenheit (°F). However, in science and engineering, where SI units are commonly used, temperature is measured in degrees Celsius (°C) as well as absolute temperature expressed in Kelvins (K).

*Heat* is defined as "thermal energy" due to the random motion of molecules.

**Temperature**, on the other hand, is a **measure** of the **concentration** of an object's "internal" thermal energy.

We can convert between the systems as follows:

$$^{\circ}F = \frac{9}{5} ^{\circ}C + 32$$
 Fahrenheit

 $^{\circ}C = \frac{5}{9} [^{\circ}F - 32]$  Celsius

 $K = ^{\circ}C + 273$  Kelvin

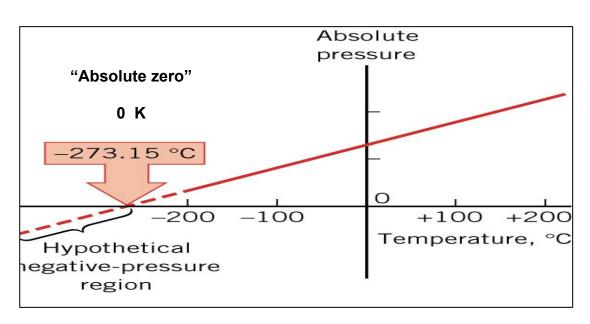
Steam point 373.15 — 100.00

Ice point 273.15 — 0.00

One kelvin equals one Celsius degree

Absolute zero 0 — - 273.15

Therefore, water *freezes* at <u>0 °C</u> [273 K] and *boils* at 100 °C [373 K]



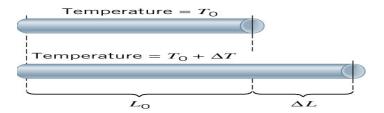
**Absolute pressure** versus temperature for a low-density gas at a constant volume.

When a *substance* undergoes a *temperature change*, it *changes* in *size*. For example, when a *metal lid* on a jar is too *tight* to open....run the lid under *hot water* to *loosen* the lid.

In other words, let the *metal expand away* from the **glass** since both solids *expand* at *different rates*.

- □ Steel beams that form railroad tracks or bridges expand when they get warmer.
- □ A balloon filled with air shrinks when placed in a freezer.

The *change in size* of a substance <u>due to</u> a *temperature change* depends on the "<u>amount</u>" of the *temperature change* and the *identity* of the *substance*.



The "identity" of a solid substance is known as it coefficient of linear expansion "alpha" ( $\alpha$ ).

## Linear Thermal Expansion of a Solid

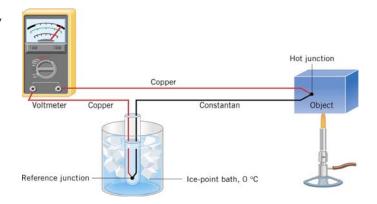
The *length*  $l_0$  of an object <u>changes</u> by an amount  $\Delta L$  when its *temperature* <u>changes</u> by an amount  $\Delta T$ .

$$\Delta L = \alpha L_0 \Delta T$$

where  $\alpha$  is the *coefficient of linear expansion* having units 1/°C or per degrees Celsius.

All thermometers make use of the change in some physical property with respect to room temperature where  $T_{\text{room}} \approx 25 \, ^{\circ}\text{C}$ . A property that changes with temperature is called a "<u>thermometric</u>" property.

A "<u>thermocouple</u>" is a thermometer used extensively in scientific laboratories. It consists of thin wires of different metals, welded together at the ends to form form two junctions.



- One junction is the "hot" junction and the other is the "reference" junction.
- □ When the thermocouple is heated, a voltage is generated based on the difference in temperature between the two junctions.
- □ With the aid of calibration tables, the temperature of the "hot" junction can be obtained by reading the voltage proportional to temperature.

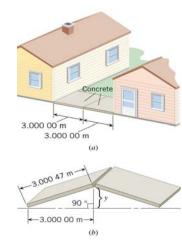
## Heat, Temperature and Linear Expansion of Solids

Ex 1: A steel beam ( $\alpha = 12 \times 10^{-6} \text{ C}^{-1}$ ) used in the construction of a bridge has a length of 30.0 m when the temperature is 15 °C. On a very hot day, when the temperature is 36 °C, how long will the beam be? [PrinctionReview9.6]

Ex 2: A cylinder of diameter 1.0 cm at 30 °C is to be slid into a hole in a steel plate ( $\alpha = 12 \times 10^{-6} \text{ C}^{-1}$ ). The hole has a diameter of 0.99970 cm at 30 °C. To what temperature must the plate be heated? [Schaums15.2]

Ex 3: Solid aluminum pipe  $(\alpha = 23 \times 10^{-6} \text{ C}^{-1})$  is used in the construction of a section of fence. By what fraction  $\Delta L/L_0$  does the length of the pipe increase when the temperature changes from  $0 \, ^{\circ}\text{C}$  on a cold winter day to 40  $^{\circ}\text{C}$  on a hot summer day? [CutnellP12.13sim]

Ex 4: A concrete sidewalk ( $\alpha = 12 \times 10^{-6} \text{ C}^{-1}$ ) is constructed between two buildings on a day when the temperature is 25 °C. The sidewalk consists of two slabs, each three meters in length and of negligible thickness. As the temperature rises to 38 °C, the slabs expand, but no space is provided for thermal expansion. The buildings do not move, so the slabs buckle upward. Determine the vertical distance y in part (b) of the drawing. [Cutnell12.3]



AP	Physics	"B"
Mr	Mirro	

Date:

Heat, Temperature and Linear Expansion of Solids

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 $\Delta L = \propto L_0 \Delta T = (12 \times 10^{-6})(36)(36-15) = 7.6 \times 10^{-3} \text{ M}$   $L' = L_0 + \Delta L = 30 + .0076 = 30.0076 \text{ m}$ 

Ex 2: A cylinder of diameter 1.0 cm at 30 °C is to be slid into a hole in a steel plate ( $\alpha = 12 \times 10^{-6} \text{ C}^{-1}$ ). The hole has a diameter of 0.99970 cm at 30 °C. To what temperature must the plate be heated? [Schaums15.2]

 $\Delta L = L_1 - L_2$   $\Delta L = 1.000 \text{ cm} - .99970 \text{ cm}$   $\Delta L = .00030 \text{ cm}$ 

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11= × lo AT So AL = × AT = (23×10 -6)(40) = (9.2×10-4)

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 $\Delta L = (12 \times 10^{-6})(3)(38-25) = .00047y$  L' = Lo + DL = 3.00 + .00097y = 3.00047y  $J = \int L'^2 - Lo^2 = \int (3.00047)^2 - (3.000)^2$   $J = \int (3.00000 \text{ m})^2$   $J = \int (3.00000 \text{ m})^2$   $J = \int (3.00000 \text{ m})^2$