

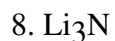
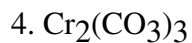
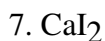
Chapter 9 Practice #2

Name _____

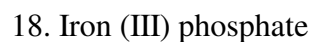
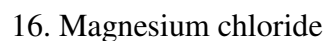
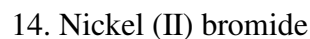
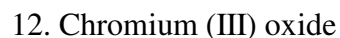
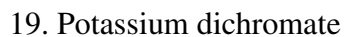
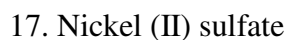
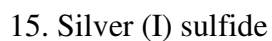
Period _____ Date _____

	Name of Cation	Name of Anion	Formula of Cation	Formula of Anion	Formula of Compound	Name of Compound
1	Calcium ion	Chloride ion				
2	Iron(III) ion	Phosphide ion				
3			Na^{1+}	S^{2-}		
4			Al^{3+}	Br^{1-}		
5						Lithium Sulfide
6						Platinum (IV) Oxide
7	Magnesium ion	Carbonate ion				
8			Ca^{2+}	NO_3^{1-}		
9					HgS	
10					$\text{Th}_3(\text{PO}_4)_2$	

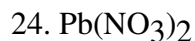
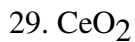
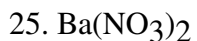
Write the names of these ionic compounds



Write the formulas for these ionic compounds



Write the names of these ionic compounds



Write the formulas of these ionic compounds.

