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**Research Article** 

# DEVELOPMENT AND VALIDATION OF ANALYTICAL METHOD FOR IRBESARTAN AND ATORVASTATIN BY SIMULTANEOUS EQUTION SPECTROSCOPIC METHOD

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#### ABSTRACT

A simple, accurate and precise spectroscopic method was developed for simultaneous estimation of Irbesartan and atorvastatin in synthetic mixture using simultaneous eqution Method.In this spectroscopic method, 226.00 nm and 246.00 nm wavelengths were selected for measurement of absorptivity. Both the drugs show linearity in a concentration range of 05-30  $\mu$ g/ml at their respective  $\lambda$ max. Accuracy, precision and recovery studies were done by QC samples covering lower, medium and high concentrations of the linearity range. The relative standard deviation for accuracy, precision studies were found to be within the acceptance range (<2%). The limit of determination was 0.033 $\mu$ g/ml and 0.125  $\mu$ g/ml for Irbesartan and atorvastatin, respectively. The limit of quantification was 0.1008  $\mu$ g/ml and 0.3792  $\mu$ g/ml for Irbesartan and atorvastatin, respectively. Recovery of Irbesartan and atorvastatin were found to be 99.75 % and 99.52% respectively confirming the accuracy of the proposed method. The proposed method is recommended for routine analysis since they are rapid, simple, accurate and also sensitive and specific by no heating and no organic solvent extraction.

Keywords: Irbesartan, atorvastatin, simultaneous estimation, Simultaneous equation method, analysis method.

## INTRODUCTION

Irbesartan, an angiotensin II receptor antagonist [1]. Is used mainly for the treatment of hypertension. It is an orally active nonpeptidetetrazole derivative and selectively inhibits angiotensin II receptor type 2. Angiotensin II receptor type1 antagonists have been widely used in treatment of diseases like hypertension, heart

Irbesartan is white or almost white, crystalline powder. Solubility is given in practically insoluble in water, sparingly soluble in methanol, slightly soluble in methylene chloride.

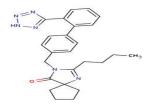


Fig.:1 Structure of Irbesartan[3]

Atorvastatin is used as lipid-lowering agents used in hyperlipidaemia condition. Atorvastatin selectively and competitively inhibits the hepatic enzyme HMG-CoA reductase. (4) As HMG-CoA reductase is responsible for converting HMG-CoA to mevalonate in the cholesterol biosynthesis pathway, this results in a subsequent decrease in hepatic cholesterol levels and decreases blood cholesterol level.

#### Atorvastatin

iswhiteoralmostwhite,crystallinepowder.Solubilityisgiveninpractical ly insoluble in water, soluble in methanol, slightlysolublein methylenechloride.

failure, myocardial infarction and diabetic nephropathy. IUPAN name of Irbesartan is 2-butyl-3-({4-[2-(2H-1,2,3,4-tetrazol-5-yl)phenyl}methyl)-1,3-diazaspiro[4.4]non-1-en-4-one.(2)

Fig. 2: Structureofatorvastatin(5)

Hypertension frequently coexists with hyperlipidaemia and both are considered to be major risk factors for developing cardiac disease ultimately resulting in adverse cardiac events. This clustering of risk factors is potentially due to a common mechanism. Further, patient compliance with the management of hypertension is generally better than patient compliance with hyperlipidaemia. It would therefore be advantageous for patients to have a single therapy which treats both of these conditions with help of fixed dose combination of Irbesartan and atorvastatin.(6,7)

The review of literature regarding quantitative analysis of Irbesartan and atorvastatinrevealed that no attempt was made to develop analytical methods for Irbesartan and atorvastatin. Some spectrometric methods and chromatographic methods have been reported for theestimation of the individual drugs. The focus of the present study was to develop and validate a rapid, stable, specific, and economic spectroscopic method for the estimation of Irbesartan and atorvastatinin Synthetic mixture. (8,9)

# MATERIALS AND METHODOLOGY

Atorvastatin and Irbesartan were obtained as gift samples from S Kant pharmaceuticals and CTX life science Surat. Synthetic Mixture contain 20mg of Atorvastatin and 160mg of Irbesartan.

- A double beam UV/Visible spectrophotometer (Shimadzu model 2450, Japan) with spectral width of 2 nm, 1 cm quartz cells was used to measure absorbance of all the solutions.
- Spectra were automatically obtained by UV-Probe system software.
- An analytical balance (Sartorius CD2250, Gottingen, Germany) was used for weighing the samples.
- Sonicator(D120/2H, TRANS-0-SONIC)
- Class 'A' volumetric glassware were used (Borosillicte)

## Standard solutionofIrbesartan (IRB)

#### Preparation of stock solution of IRB

Accurately weighed quantity of Irbesartan 10 mg was transferred to 100 ml volumetric flask, dissolved and diluted up to mark with methanol to give a stock solution having strength of  $100\mu g/ml$ .

# Preparation of stock solution of ATR

Accurately weighed quantity of Atorvastatin 10mg was transferred to 100 ml volumetric flask, dissolved and diluted up to mark with methanol to give a stock solution having strength of  $100\mu g/ml$ .

#### Preparation of standard mixture solution

From the stock solution of IRB take 1.6ml and from stock solution of ATR take 0.2ml and transferred in to 10ml volumetric flask and diluted up to mark with methanol to give a solution having strength of IRB was  $16~\mu g/ml$  and ATR was  $2\mu g/ml$ .

#### Preparation of test solution

From the stock solution of IRB take 1.6ml and from stock solution of ATR take 0.2ml and transferred in to 10ml volumetric flask and diluted up to mark with methanol to give a solution having strength of IRB was  $16 \,\mu\text{g/ml}$  and ATR was  $2\mu\text{g/ml}$ .

#### alibration curves for Irbesartan

Pipette out 0.5, 1.0, 1.5, 2.0, 2.5 and 3.0 ml of the stock solution of Irbesartan and atorvastatin ( $100\mu g/ml$ ) into a series of 10ml volumetric flasks and the volume was adjusted to mark with methanol and measured absorbance at 226.00nm and 246nm. Plotte the graph of absorbance versus respective concentration of Irbesartan and atorvastatin. Linearityrange of IRB and ATR was found with correlation co-efficient.

# DEVELOPMENT AND VALIDATION OF SPECTROSCOPIC SIMULTANEOUS EQUATION METHOD

# SELECTION OFWAVELENGTHAND METHOD DEVELOPMENT FORDETERMINATION OF IRBESARTAN AND ATORVASTATIN

The standard solution of IRB and ATR were scanned separately between 200-400 nm, and IRB showed absorbance maxima at 226.00 nm and ATR at 246.00 nm. (figure 3)

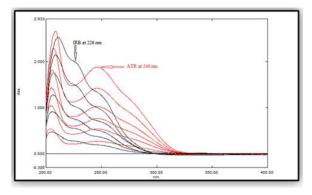


Fig.3 Overlainzero orderspectra ofIRB andATR(8:1) ratios, respectively

# Table 1Calibrationdata for IRB and ATRat 226.00 nm and 246.00 nm respectively. \*(n=6)

Sr. No	Concentration (μg/ml)		Absorbance* (226.00nm)±SD IRB	Absorbance* (246.00nm)±SD ATR	
1 2	IRB 05 10	<b>ATR</b> 05 10	0.3708±0.0023 0.7460±0.0020	0.2672±0.0015 0.5674±0.0017	
3	15	15	1.2171±0.0013	0.8872±0.0018	
4	20	20	1.6972±0.0015	1.1974±0.0012	
5	25	25	2.2225±0.0013	1.5232±0.0022	
6	30	30	2.7653±0.0025	1.8772±0.0016	

# **VALIDATION PARAMETERS(10)**

#### Linearity andRange

The Zero order (fig. 3) showed linear absorbance at 226.00 n m for IRB (05-30  $\mu$ g/ml) and 246.00 n m for ATR (5-30

 $\mbox{\sc \mu g/ml}\mbox{\sc with correlation coefficient} (\mbox{\sc r}^2) of 0.9994 and 0.9993 for IRB and ATR, respectively.$ 

Thismethodobeyedbeer's law in the concentration range 05  $30\,\mu g/ml$  and 5  $-30\,\mu g/ml$  for IRB and ATR, respectively. (Table 1)

Correlationcoefficient(r<sup>2</sup>)forcalibrationcurveofIRBandATRwasfoun d to be 0.9994 and 0.9993,respectively(figure4and 5)

Theregression line equation for IRB and ATR are as following,

v = 0.0983x - 0.2385 for IRB (1

y = 0.0642x - 0.0695 for ATR \_\_\_\_\_(2)

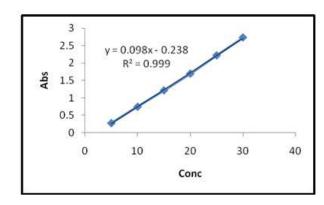


Fig.4CalibrationcurveforIRB at 226.00 nm

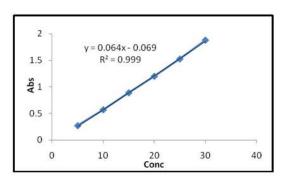


Fig. 5CalibrationcurveforATR at 246.00 nm

#### Precision

## **Intraday precision**

The precision of the developed method was assessed by analyzing combined standard solution containing three different concentrations 05, 15, 30  $\mu g/ml$  for IRB and 05, 15, 30  $\mu g/ml$  ATR. Three replicate (n=3) each on same day. Intraday precision data presented in Table 2

#### These%

 $RSD value was found to be less than {\pm} 2.0 indicated that the method is precise. \\$ 

#### Table 2 Intraday precision data for estimation of IRB and ATR\*(n=3)

Conc. (µg/ml)		IRB Abs.* ± % RSD ±% RSD Abs. ±% RSD IRB	ATR Abs.*± % RSD
IRB	ATR		
05	05	0.372±0.45	0.266±0.57
15	15	1.211±0.21	0.884±0.92
30	30	2.763±0.52	1.877±0.23

#### Interdayprecision

The precision of the developed method was assessed by analyzing combined standard solution containing three different

concentrations 05, 15, 30  $\mu$ g/ml for IRB and 05, 15, 30  $\mu$ g/ml ATR triplicate (n=3) per day for consecutive 3 days for inter-day precision. Interday precision data presented in Table 3

 $The se\%RSD value was found to be less than \pm 2.0 indicated that the method is precise.\\$ 

#### Table 3Interdayprecision data for estimation of IRB and ATR\*(n=3)

Conc. (µg/ml)		IRB Abs.* ±% RSD	ATR Abs.*±%RSD
		±% RSD Abs.	
IRB	ATR	±% RSD IRB	
05	05	0.377±0.55	0.270±0.56
15	15	1.215±0.25	0.887±0.17
30	30	2.786±0.85	1.881±0.36

#### Accuracy

Accuracyofthemethodwasdeterminedbyrecoverystudyfromsynthe tic mixture at threelevel (80%, 100%, 120%)of standard addition. The% recoveryvalues are tabulated in Table 4 and 5

Percentage recovery for IRB and ATR by this method was found in the range of 100.07 to 100.43% and 99.21 to 100.55%, respectively,

The value of % RSD within the limit indicated that the method is accurate and percentage recovery shows that there is no interference from the excepients.

#### Table 4Recovery data ofIRB\*(n=3)

Conc. ofIRB from formulation (µg/ml)	Amount of Std.IRB added (µg/ml)	Total amount of IRB (μg/ml)	Total amount ofIRB found (µg/ml)* Mean± SD	% Recovery (n=3)	% RSD IRB
8	6.4	14.4	12.81±0.022	100.07%	0.32%
8	8.0	16.6	16.07±0.013	100.43%	0.68%
_ 8	9.6	17.6	19.22±0.045	100.10%	0.28%

## Table 5Recovery data of ATR\*(n=3)

Conc. ofATR from formulation (μg/ml)	Amount of Std.ATR added (µg/ml)	Total amount of ATR (µg/ml)	Total amount ofATR found (μg/ml)* Mean± SD	% Recovery (n=3)	% RSD ATR
1	0.8	1.8	1.81±0.021	100.55%	0.84%
1	1.0	2.0	2.00±0.036	100.50%	0.22%
1	1.2	2.2	2.19±0.20	99.21%	0.35%

/ml, respectively.

μg/ml, respectively.

 $The LOQ for IRB and ATR was conformed to be 0.1008 \mu g/ml and 0.379$ 

 $The obtained LOD and LOQ results\ are presented in\ Table\ 6$ 

## Table 6LOD andLOQ dataofIRB andATR \*(n=10)

Conc. (	(μg/ml)	Avg.abs* ± SD (226.00nm) IRB	% RSD	Avg.abs*±SD (246.00nm) ATR	% RSD
IRB	ATR	,		ATR	
5	5	0.371 ±0.0007	1.93	0.270 ±0.0006	0.45
LOD (	μg/ml)	0.033		0.125	
	μg/ml)	0.1008		0.3792	

#### **Robustness and Ruggedness**

The obtained Ruggedness and Robustness results are presented in table 6.3.8

The % R.S.D was found to be 0.12 – 0.84 % for IRB and 0.11 – 0.74 % for ATR.

These %RSD value was found to be less than ± 2.0 indicated that the method is precise.

No significant changes in the spectrums were observed, proving that the developed method is rugged and robust.

# Table 7 Robustness and Ruggedness data of IRB and ATR \*(n=3)

Condition	Conc. (µg/ml)	Different Instrument		•	Different Stock Solution eparation
		UV-2450	UV-1800	Stock-1*	Stock-2*
Irbesartan	05	0.376±0.32	0.374±0.47	0.376±0.12	0.373±0.82
Mean (n=3)	15	1.215±0.56	1.216±0.22	1.215±0.42	1.216±0.56
± % RSD	30	2.763±0.23	2.765±0.84	2.764±0.21	2.763±0.32
Atorvastatin	05	0.271±0.54	0.269±0.43	0.272±0.42	0.270±0.11
Mean(n=3)	15	0.885±0.66	0.882±0.33	0.884±0.15	0.885±0.33
± %RSD	30	1.879±0.16	1.878±0.13	1.882±0.52	1.884±0.74

Stock-1:- 10 mg dissolve in 100 ml Methanol

Stock-2:- 50 mg dissolve in 250 ml Methanol

# APPLICATION OF THE PROPOSED METHOD FOR ANALYSIS OF IRB AND ATR IN COMBINED CAPSULE DOSAGE FORM.

All the excipients were mixed in 10ml volumetric flask and sonicate for 15min. make up the volume with Distilled Water. The solution was filtered through Whatman filter paper No. 42.

Finally the solution had concentration  $1600\mu g/ml$  for IRB and  $200\mu g/ml$  for ATR. from that pipette out 0.1ml in 10 ml volumetric flask and volumewasmadeuptomarkwithmethanol

toobtainfinal solutioncontaining 16  $\mu g/mlof IRB~$  and  $~2\mu g/ml~$  of ATR. Azeroorder ofthe resulting spectrum solution was recorded and processed to first derivative spectra. As pectrumofthesamplesolution wasrecordedandtheabsorbanceat 226.00nmand246.00nmwerenoted forestimationofIRB andATR,respectively.The concentrations of IRB and ATR in formulation determinedusing thecorresponding were calibrationgraph.

# Table 8 Analysisdata of commercial formulation\*(n=3)

Sr. N	o Drug	Formulation (µg /ml)	% Assay* ± SD	USP limit(%)
1	IRB	16.0	99.75 ± 0.22	98-102%
2	ATR	2.0	99.52 ± 0.56	98-102%

#### SUMMARY OF VALIDATION PARAMETER

# Table 9 Summary of validation parameters

SR. NO.	PARAMETER	Irbesartan	Atorvastatin	
1	Wave length Max.	226.00 nm	246.00 nm	
2	Linearity (µg/ml) (n=6)	5 to 30 μg/ml	5 to 30 μg/ml	
3	Regression equation	y = 0.0983x - 0.2385	y = 0.0642x - 0.0695	
4	Correlation coefficient (r <sup>2</sup> )	0.9994	0.9993	
5	Accuracy(%Recovery) (n=3)	100.26	100.13	
6	Precision			
	Intra-day (%RSD)(n=3)	0.21-0.52	0.23-0.92	
	Inter-day (%RSD)(n=3)	0.25-0.85	0.17-0.56	
7	LOD ( $\mu$ g/ml) (n=10)	0.033	0.125	
8	LOO (ug/ml) (n=10)	0.1008	0.3792	

9	Robustness and Ruggedness (%RSD)	0.12-0.84	0.11-0.73	
10	Assay	99.75±0.22	99.52 ±0.56	

#### CONCLUSION

A new, Simultaneous Equation method has been developed for estimation of Irbesartan and Atorvastatin in synthetic mixture. The method was validated by employment of ICH(18) guidelines. The validation data is indicative of good precision and accuracy, and prove the reliability of the method.

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