

BUREAU INTERNATIONAL DES POIDS ET MESURES

**International comparison of activity measurements of a solution of ^3H
(January 2009)**

Participating laboratory:

 $T_{1/2} = (4\,496.862 \text{ d}; u = 9.131 \text{ d})^*$

Ampoule number _ _ _ _ _

Mass of solution, according to distributing laboratory _ _ _ _ _ g

Name(s) of the person(s) who carried out the measurements:

Date:

Please send the filled-in form and any additional information to the BIPM
*not later than **May 31, 2009**.*

* M.M. Bé et al., [*Monographie BIPM-5, Table of radionuclides, Vol. 3*](#), 2006, 210 pp, p. 1

A. Preliminary measurements

A.1. Method used for preliminary measurements

- | | | | | |
|---------------------------------|-----|--------------------------|----|--------------------------|
| - calibrated ionization chamber | YES | <input type="checkbox"/> | NO | <input type="checkbox"/> |
| - well crystal | YES | <input type="checkbox"/> | NO | <input type="checkbox"/> |
| - other method | YES | <input type="checkbox"/> | NO | <input type="checkbox"/> |

A.2. Results obtained

Radioactivity concentration, in kBq g⁻¹ (2009-05-31, 0 h UTC)

- before opening the original ampoule _____
 date of this measurement _____
- after transfer to another ampoule _____
 date of this measurement _____

Total mass of solution found in the ampoule _____ g

A.3. Adsorption tests

Please take into account the adsorption tests in the evaluation of the final results.

A.3.1. Adsorption tests carried out with liquid-scintillation counting

- Count-rate after two rinsings: _____ s⁻¹
(converted to activity by applying the measured detection efficiency).

Thus, activity remaining in the “empty” original ampoule : _____ Bq.

Date of this test: _____

Number of additional rinsings _____

Final residual count-rate: _____

Thus, final residual activity:

Date of final test measurements: _____

Please explain the measuring procedure used:

A.3.2. Adsorption tests carried out with proportional counting

Please rinse the ampoule with an aggressive solution to remove most of the activity and prepare solid source(s) to measure this residual activity.

Activity remaining in the “empty” original _____ Bq.

Date of this test _____.

A.4. Impurity checks* :

Method of measurement _____,

Nuclide _____,

Impurity to ^3H ratio _____ and its uncertainty _____

at reference date (2009-05-31, 0 h UTC).

Please note that any impurity determination is considered to be an important part of the comparison.

* Please give this information for each impurity found.

B. Source preparation

B.1. Methods used for source preparation:

Possible remarks about drying, precipitation, foils used (gold-coated or not, number, etc.), type of balance used...

B.2. Solutions, sources

B.2.1. *For photon counting and electron counting (if relevant)*

Diluent:

	<i>d i l u t i o n n u m b e r</i>			
	1	2	3	...
- dilution factor	-----	-----	-----	
possible remarks	-----	-----	-----	
- number of sources prepared	-----	-----	-----	
- disposed mass of solution (approx.)	-----	-----	-----	

B.2.2. *For liquid-scintillation counting*

Diluent	-----	
Dilution factor	-----	
Scintillator used to prepare the sources	-----	
Volume of scintillator used	-----	cm ³
Chemicals used to stabilize the solution	-----	
Substances used as quenching agent	-----	
Type of vials used	-----	

C. Procedures used for the activity measurements

Method of measurement used

(e.g. efficiency tracing method with a pressurized proportional counter, efficiency tracing method with a liquid-scintillation spectrometer, CIEMAT/NIST or TDCR method or by means of any other experimental devices)

Please list the values for all the decay-scheme parameters (branching ratios, transition intensities, internal conversion coefficients, etc.) relevant to your measurements.

In case you used more than one method, please assemble the relevant information on separate sheets.

D. Detectors, counting equipment

D.1. Photon counting

D.1.1. Scintillator detector

Crystal material	_____	Solid angle	_____	sr
Number of crystals	_____	Well type	YES <input type="checkbox"/> NO <input type="checkbox"/>	
Crystal diameter	_____ mm	Crystal height	_____	mm
Well diameter	_____ mm	Well depth	_____	mm
Window material	_____	Thickness	_____	mm
Distance between photon counter and source			_____	mm
Resolution at	_____ keV,	FWHM*	_____ %,	_____ keV

Please add a typical pulse-height spectrum.

D.1.2. Semiconductor detector

Nature	_____	Solid angle	_____	sr
Number of detectors	_____	Well type	YES <input type="checkbox"/> NO <input type="checkbox"/>	
Type	_____	Coaxial	<input type="checkbox"/> Planar <input type="checkbox"/>	
Diameter	_____ mm	Volume	_____	cm ³
Window material	_____	Window thickness	_____	mm
Distance between photon counter and source			_____	mm
Resolution at	_____ keV,	FWHM*	_____ %,	_____ keV

Please add a typical pulse-height spectrum and an efficiency curve.

* full width at half maximum

Radionuclides used for an efficiency determination (if relevant)

P_γ (%)

-----	-----
-----	-----
-----	-----
-----	-----
-----	-----
-----	-----

D.1.3. *Other detectors used*

D.2. Photon and/or electron counting

D.2.1. *Proportional counter*

Solid angle ----- sr

Wall material ----- Height of each half ----- mm

Anode

Nature -----

Wire diameter ----- mm Wire length ----- mm

Distance from source ----- mm

Voltage applied ----- kV

Gas

Nature -----

Pressure

(above atmospheric pressure) ----- MPa

Remarks

D.2.2. *Liquid-scintillation equipment*

D.2.2.1. *CIEMAT/NIST method*

D.2.2.1.1. *Characterization of the liquid-scintillation counter (LSC)*

Type of the counter	-----
Age	-----
Quench parameter	-----
Nuclide used as external standard	-----
Efficiency obtained with an unquenched standard of ^3H	-----
Background (unquenched standard in toluene scintillator, 0 to 2000 or more keV)	-----
Options used (e.g. low-level counting)	-----

D.2.2.1.2. *Characterization of the tracer (e. g. ^{54}Mn)*

Standard used and its origin	-----
Uncertainty on the standard	-----
Date of preparation of the tracer samples	-----
Chemical composition of the tracer samples	-----

D.2.2.2. *TDCCR method*

D.2.2.2.1. *Characteristics of the experimental equipment*

Type of phototubes	-----
Operating temperature	-----
Coincidence resolving time	-----

Type of dead time	extending <input type="checkbox"/>	non-extending <input type="checkbox"/>
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Minimum dead-time length	----- μs
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Efficiency variation method:

- defocusing	-----
- grey filters	-----
- chemical quenching	-----
- other ones (please describe)	-----

External standard (^3H or other) used for the determination of the figure of merit	-----
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D.2.2.3. *Other method(s)*

(For other detector or additional channels, please give information on appropriate sheets.)

D.4. Parameters of counting equipment

(Give a brief description, e.g. in the form of a block diagram.)

D.4.1. *Photon counting*

- a) Discrimination level _____ keV
(or window)
- b) Dead times and their uncertainties (standard deviation)

Dead time $\tau_1 = \text{_____ } \mu\text{s}; u = \text{_____ } \mu\text{s}$

Type of dead time extending ☐ non-extending ☐

Method used for measurement _____

D.4.2. *Electron counting*

- a) Discrimination level _____ keV
(or window)
- b) Dead times and their uncertainties (standard deviation)

Dead time $\tau_2 = \text{_____ } \mu\text{s}; u = \text{_____ } \mu\text{s}$

Type of dead time extending ☐ non-extending ☐

Method used for measurement _____

D.4.3. *Coincidence unit (if relevant)*

Coincidence resolving time $\tau_R = \text{_____ } \mu\text{s}; u = \text{_____ } \mu\text{s}$

Method used for measurement

D.4.4. *Other modules used*

(for LSC see section D.2.2.)

E. Relevant data, corrections and uncertainties

E.1. Date of measurement of ^3H

(Mean date on which your measurements were carried out)

E.2. Measuring data

E.2.1. *Photons*

Dead time	_____ μs	Number of sources measured	_____
Background rate	_____ s^{-1}	Typical count rate	_____ s^{-1}
Typical time for one measurement	_____ s		
Discrimination threshold or window	_____ keV		

E.2.2. *Electrons*

Dead time	_____ μs	Number of sources measured	_____
Background rate	_____ s^{-1}	Typical count rate	_____ s^{-1}
Typical time for one measurement	_____ s		
Discrimination threshold or window	_____ keV		

E.2.3. *Extrapolation of efficiency function*

Maximum achieved efficiency	_____ %
Method used for varying the efficiency	

Number of degrees of freedom _____

Please add a figure, if possible.

E.2.4. *Liquid-scintillation parameters*

Numerical codes used	-----	
kB value	-----	unit -----
Formula used to calculate the ionization quenching correction factor $Q(E)$ -----		

Are M, N, ... captures taken into account?		-----
Are M, N, ... x-ray and Auger electrons taken into account?		-----
Model used to evaluate the interaction probability of		
the photons with the scintillator -----		
Values used for cross section of interaction		-----

E.2.5. *Calculated data for the liquid-scintillation method*

Total efficiency ^3H -----

E.2.6. *Corrections applied*

E.2.7 *Uncertainty components**, in % of the activity concentration, due to

		Sensitivity factors	Type A or B method	Remarks
counting statistics	-----	-----	-----	-----
weighing	-----	-----	-----	-----
dead time	-----	-----	-----	-----
background	-----	-----	-----	-----
pile-up	-----	-----	-----	-----
counting time	-----	-----	-----	-----
adsorption	-----	-----	-----	-----
impurities	-----	-----	-----	-----
tracer	-----	-----	-----	-----
input parameters and statistical model	-----	-----	-----	-----
quenching	-----	-----	-----	-----
interpolation from calibration curve	-----	-----	-----	-----
decay-scheme parameters	-----	-----	-----	-----
half life ($T_{1/2} = 4\,496.862\text{ d}$; $u = 9.131\text{ d}$)	-----	-----	-----	-----
self absorption	-----	-----	-----	-----
extrapolation of efficiency curve	-----	-----	-----	-----
other effects (if relevant) (explain)	-----	-----	-----	-----
combined uncertainty (as quadratic sum of all uncertainty components)	-----	-----	-----	-----

* The uncertainty components are to be considered as approximations of the corresponding standard deviations (see also *Metrologia*, 1981, **17**, 73 and *Guide to expression of uncertainty in measurement*, ISO, corrected and reprinted 1995).

F. Combination of individual results

(obtained from the individual dilutions, source preparation, etc.)

How have the individual results been used for arriving at the final result (statistical weights)?

G. Final result for a given method

(Please, for each method used, give the following information)

The radioactivity concentration of the ^3H solution on the reference date
(2009-05-31, 0 h UTC) is

_____ kBq g^{-1} ,

and the combined uncertainty is

_____ kBq g^{-1} , _____ %

Remarks

H. Laboratory final result

If several measuring methods were used, please indicate below the value to be considered for the evaluation of the degree of equivalence (according to the CIPM MRA), and its method of calculation.

The radioactivity concentration of the ^3H solution on the reference date (2009-05-31, 0 h UTC) is

_____ kBq g^{-1} ,

and the combined uncertainty is

_____ kBq g^{-1} , _____ %

Remarks