| Name | Date | Class |
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Elements and the Periodic Table • Review and Reinforce

## **Metals**

## **Understanding Main Ideas**

Answer the following on a separate sheet of paper. Use the periodic table in Appendix D.

- **1.** What category of element is the most common in the periodic table? Where is that category found in the periodic table?
- **2.** Sodium (Na) and calcium (Ca) are in different families of metals. Name the families of metals in which they belong, and describe each family's characteristics.
- **3.** Would a metal in Group 13 be more or less reactive than a metal in Group 1? Explain.
- **4.** In what periods are the lanthanides and actinides? Where are they placed in the periodic table? Why?

## **Building Vocabulary**

Write the correct term on each line to complete the sentence.

conductivity
alloy
malleable
ductile
corrosion
particle accelerator
reactivity

| rea | nvity  |
|-----|--|
| 5.  | Γhe reaction of a metal with oxygen to form rust is called   |
| 6.  | A material that is can be hammered into sheets and other shapes.   |
| 7.  | A powerful machine called a(n) moves nuclear particles fast enough to make larger nuclei when the particles collide. |
| 8.  | The ability to transmit heat or electricity to other objects is called   |
| 9.  | A material that is can be drawn into a wire.   |
| 10. | is the ease and speed with which an element  |
|     | combines with other elements and compounds.  |
| 11. | A(n) is a mixture of a metal and at least one  |
|     | other element.   |