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General

- □ If a **footnote management system** such as EndNote is used for the references, please clear all fields from the references using [ctrl]+[shift]+F9.
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- In the Experimental Section, quantities of reactants, solvents etc. should be included in parentheses rather than in the running text
 e.g., triphenylstannyl chloride (0.964 g, 2.5 mmol) in toluene (20 mL).
- □ Data should be quoted with decimal points (not commas) and negative exponents (for example: 25.8 J K⁻¹ mol⁻¹). Characterization format should be: mp/bp 20 °C; $[a]_D^{20} = -13.5$ (*c* = 0.2, in acetone); ¹H NMR (400 MHz, DMSO-*d*₆, δ): 7.15 (s, 2H, Ar H), 1.3 (q, *J* = 8 Hz, 2H; CH₂), 0.9 (t, *J* = 8 Hz, 3H; CH₃); ¹³C NMR (100 MHz, CDCl₃, δ): 175.4 (C=O), 156.5 (C4); IR (KBr): *v* = 2972 (w), 2907 (w), ..., 1026 (s; *v*_{as}(SiOSi)), 971 (*v*_s), ..., 666 (w; *v*_s(SiOSi)), ..., 439 (m), 401 cm⁻¹ (m); UV-vis (*n*-hexane): $\lambda_{max} (\varepsilon) = 320$ (5000), 270 nm (12000); EIMS *m*/*z* (%): 108 (20) [M⁺], 107 (60) [M⁺ H], 91 (100) [C₇H₇⁺]; HRMS (ESI) *m*/*z*: [M + H]⁺ calcd for C₂₁H₃₈N₄O₆S, 475.2591; found, 475.2593. Anal. calcd for C₄₅H₂₈N₄O₇: C 62.47, H 3.41, N 6.78; found: C 62.27, H 3.46, N 6.80.
- □ Abbreviations and acronyms should be used sparingly and consistently. Where they first appear in the text, the complete term in lower case—apart from the most common ones (see the list on the journal homepage)—should also be given. In articles where a large number of abbreviations and acronyms occur, these can be explained in a glossary at the end.

Literature Section

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- [1] a) H. J. Ache, Angew. Chem. 1989, 101, 1; Angew. Chem. Int. Ed. Engl. 1989, 28, 1; b) S. C. Stinson, Chem. Eng.

News 1998, 76, 57; c) A. Jones, B. Smith, Adv. Mater., in press; d) A. Jones, B. Meyer, Macromol. Rapid Commun.,

DOI: 10.1002/marc.203001234.

[2] H. R. Allcock, Introduction to Materials Chemistry, Wiley, Hoboken, NJ, USA 2008.

[3] J. W. Grate, G. C. Frye, in Sensors Update, Vol. 2 (Eds: H. Baltes, W. Göpel, J. Hesse), Wiley-VCH, Weinheim,

Germany 1996, Ch. 2.

[4] a) A. Katschalsky, A. Oplatka, in Proc. of the 4th Int. Conf. on Rheology (Eds: E. H. Lee, A. L. Copley),

Interscience, New York 1965; b) A. Katschalsky, presented at the 4th Int. Conf. on Rheology, San Diego, CA,

September 1964.

[5] a) A. Smith, Ph.D. Thesis, University of Newcastle, UK 1991; b) W. Lehmann, H. Rinke (Bayer AG), Ger. 838217,

1952.

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